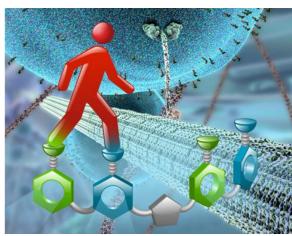
Giving Chemistry Direction

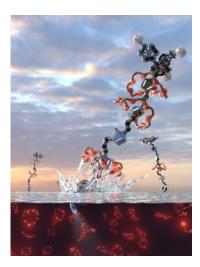
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In recent years examples of synthetic molecular machines and motors¹ have been developed,² all be they primitive by biological standards. Such molecules are best designed to work through statistical mechanisms. In a manner reminiscent of Maxwell's Demon,³ random thermal motion is rectified through ratchet mechanisms³⁻⁷ giving chemistry direction.

It is increasingly being recognised that similar concepts can be applied to other chemical exchange processes.⁸ Ratchet mechanisms—effectively chemical engines⁹ in which catalysis^{4,6,7} of 'fuel' to 'waste' drives another chemical process—cause directional impetus in what are otherwise stochastic systems, including reversible chemical reactions. This is ushering in a new era of non-equilibrium chemistry, providing fundamental advances in functional molecule design, overturning existing dogma and offering fresh insights into biology and molecular nanotechnology.







For a musical introduction, see 'Nanobot': https://bit.ly/2M5Zwdl

[1] The Nobel Prize in Chemistry 2016—Advanced Information. Nobelprize.org. Nobel Media AB 2014. Web. 6 Oct, 2016, http://www.nobelprize.org/nobel-prizes/chemistry/laureates/2016/advanced.html. [2] "Rise of the molecular machines", Angew. Chem. Int. Ed. 54, 10080 (2015). [3] "A molecular information ratchet", Nature 445, 523 (2007). [4] "An autonomous chemically fuelled small-molecule motor", Nature 534, 235 (2016). [5] "Rotary and linear molecular motors driven by pulses of a chemical fuel", Science 358, 340 (2017). [6] "A catalysis-driven artificial molecular pump", Nature 594, 529 (2021). [7] "Autonomous fuelled directional rotation about a covalent single bond", Nature 604, 80 (2022). [8] "Design, synthesis and operation of small molecules that walk along tracks", J. Am. Chem. Soc. 132, 16134 (2010). [9] "Chemical engines: Driving systems away from equilibrium through catalyst reaction cycles", Nat. Nanotechnol. 16, 1057 (2021).